

Engineering And Chemical Thermodynamics Koretsky Solutions

Volatility (chemistry) 279–281. 639–641. ISBN 978-1-119-17764-7. A highly volatile substance such as rubbing alcohol (isopropyl alcohol) will quickly evaporate, while a substance with low volatility such as vegetable oil will remain. Zumdahl, In chemistry, volatility is a material quality which describes how readily a substance vaporizes. At a given temperature and pressure, a substance with high volatility is more likely to exist as a vapour, while a substance with low volatility is more likely to be a liquid or solid. Sons. Koretsky, Milo D. Differences in volatility can be observed by comparing how fast substances within a group evaporate (or sublime in the case of solids) when exposed to the atmosphere. (2013). John Wiley & Sons. Volatility can also describe the tendency of a vapor to condense into a liquid or solid; less volatile substances will more readily condense from a vapor than highly volatile ones. pp. Engineering and Chemical Thermodynamics. pp 608c3fd4d6

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